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portion of the Study Guide to review what you have read in every section of the textbook and to

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building blocks of proteins. An example of a polysaccharide is. cellulose is a polymer made of. building blocks of carbohydrates. amino

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acid. glycogen.
glucose.
monosaccharides
(simple sugars)

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JUNCTION**

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nucleus atom's central
core. proton positively
charged, found in the
atomic nucleus neutron
no electrical charge,
found in the atomic
nucleus electron
negatively charged,
surround the nucleus in

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a "cloud" ; orbit the nucleus Can only have 2 electron charges on

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the building blocks of
matter. identify the

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building blocks of cells.
identify the 3
subatomic particles of
a.... using magnesium
as an example,
illustra.... atoms.
carbohydrates,
proteins, lipids, nucleic
acids (not sure abt t....

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Basic unit of chapter 2
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2 Study Guide The

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Biology.; what kind of bond shares electrons?
Study A Wasserman,
study 2 acetylcholine
and cholinesterase do?
2 order biology
continue enjoying
guide the, guide Away
Service, chapter that
takes up chapter and
has mass.

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from BIO 103 at
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1) List the
commonalities shared
by living organisms. a.
Require liquid water b.
Have a common set of
biological

Biology Chapter 2 Study Guide - 1 List the commonalities

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In Chapter 2, you have
learned that water is a

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highly polar molecule that forms hydrogen bonds with other water molecules. Many of the physical properties of water—such as surface tension, capillarity, adhesion, and cohesion—result from hydrogen bonding.

INTERMOLECULAR FORCES CAUSE SURFACE TENSION.

SECTION ATOMS, IONS, AND MOLECULES 2.1

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Life is beautiful! From atoms to cells, from genes to proteins, from populations to ecosystems, biology is the study of the fascinating and intricate systems that make life possible.

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Chapter 2 Biology Study Guide. atom's central core. negatively charged, surround the nucleus in a "cloud" ; orbit the nucleus Can only have 2 electron charges on shell. - orbit the nucleus of an atom in specific electron shells. - number of _____ in the

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outermost shell
determines the
chemical properties of
an atom.

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Chemistry Of Life Study
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such as bread, fruit,
and cheese are rich
sources of biological
macromolecules. The
elements carbon,
hydrogen, nitrogen,
oxygen, sulfur, and
phosphorus are the key
building blocks of the

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chemicals found in living things.

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Littel pages 34-64.

Below you find the classroom assignments and PPT's used for Chapter 2, The Chemistry of Life. You may use this website for access to PPT's, guided notes, and make up assignments. ... Chapter 2 Chemistry of Life Study Guide

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