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particle not directly involved in a chemical reaction. combustion reaction. a reaction in which oxygen reacts with another substance, often producing light or heat. reactant.

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space provided, write the letter of the
term or phrase that best completes each
statement or best answers each
question. _____ 1. You mix solution A
with solution B in a

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Chapter Assessment Chemical Reactions Answers

Chapter 7 Pretest: Chemical Reactions
Multiple Choice Identify the choice that
best completes the statement or
answers the question. ____ 1. When a
chemical change occurs: a. atoms are
rearranged. b. the law of conservation of
mass is always obeyed. c. the chemical

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properties of new substances are different from the ones you started with.
d.

Chapter 7 Pretest: Chemical Reactions - School Specialty

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chemical properties of that element is
a(n) ... simple whole-number ratios to
form chemical compounds. In chemical
reactions, atoms are combined,
separated ...

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Reactions In the space provided, write
the letter of the term or phrase that best
completes each statement or best
answers each question. _____ 1. You mix
solution A with solution B in a beaker.
Which of the following observations does

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not help you prove that a chemical reaction ...

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taken place in which of the following
process? (i) Ice melts into water. (ii) A
wet shirt got dried in sunlight. (iii) A
brown layer is formed over iron rod kept

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in air. (iv) Sugar getting dissolved in water.

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Tests Chemical Reactions (continued)
32. A) Enzymes are catalysts. They
speed up chemical reactions by lowering
the activation energy of reactions
needed for life. Enzymes allow reactions
that are necessary for life to occur at

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body temperature. B) Sawdust has a greater surface area than wood.
Increased

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magnesium and oxygen in MgO. Mg**
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Reactions PART I In the space provided,
write the letter of the term or phrase
that best completes each statement or
best answers each question. _____ 1. The
production of a slightly soluble solid

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compound in a double-displacement reaction results in the formation of a a. gas. b ...

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