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SECTION 1 SHORT

ANSWER 1. c 2. a 3. b

2. a. alcohol b. water c.

the gels 3. The mixture

is a colloid. The

properties are

consistent with those

reported in Table 3 on

page 404 of the text.

The particle size is

small, but not too

small, and the mixture

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bond. bond energy.
ionic bond. covalent
bond. a force that
holds groups of two or
more atoms together
and ma.... the energy
required to break the
bond. results from
attractions among
closely packed
oppositely charg....
results when similar or
identical atoms are
brought close toge....

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SECTION 1 SHORT

ANSWER 1 c 2 a 3 b 2 a

alcohol b water c the

gels 3 The mixture is a

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12 Answers - Zumdahl
12.19 The coefficients
in the balanced
reaction relate the rate
of disappearance of
reactants to the rate of
production of products.
From the balanced
reaction, the rate of
production of P₄ will
be $\frac{1}{4}$ the rate of

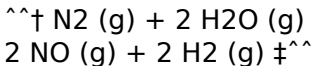
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disappearance of PH_3 ,
and the rate of
production of H_2 will
be $\frac{6}{4}$ the rate of

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Answers shown below:



At equilibrium $[\text{NO}] =$
 0.062 M . Calculate the
equilibrium constant,

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CK-12 Chemistry.

Answer Key 1.c 2.a 3.c

4.b 5.d 6.false

7.macroscopic 8.true

9.false 10.true 11.false

12.applied 13.pure

14.c 15. a.the initial

research was most

likely a pure chemistry

type of project,

learning how to make

polymers and studying

the properties of these

reactions. b.the applied

chemistry came in

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when they wanted to
make a specific
product with ...

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Chapter 12 Review,
Mixed Review:
Solutions Although
there are only six
questions on this
chemistry handout, it
makes a thorough
review of solutions.
Novices explain why a
compound is not an

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electrolyte, identify types of compounds, and calculate moles, grams, and molalities in various solutions.

Chemistry Chapter 12 Mixed Review Answers

CHAPTER 12 REVIEW
Solutions SECTION 1
SHORT ANSWER

Answer the following questions in the space provided. 1. Match the type of mixture on the left to its

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representative particle diameter on the right.

c solutions (a) larger than 1000 nm a

suspensions (b) 1 nm to 1000 nm b colloids

(c) smaller than 1 nm

2. Identify the solvent in each of the following examples:

mc06se cFMsr i-vi

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Basics of Chemistry

Deck (33) 0 If a

product has a pH of
8.9, it is considered

___ . Alkaline 1 There
are about ___

naturally occurring
elements, each with its
own distinctive
physical and chemical
properties. 90 2

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part 2. 79 terms.

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terms. DBMS VOCAB.

36 terms. Chapter
9/12. 15 terms.

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Review Solutions

Section 2 Answers.

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CHAPTER 1 REVIEW.

Matter and Change.

SECTION 1. SHORT

ANSWER Answer the

..... 12. 18. B. 18. 12. 6.

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CHAPTER 3 REVIEW

Atoms: The Building

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Blocks of Matter MIXED
REVIEW SHORT

ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

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Winston Modern
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choice exercise.

Choose the correct

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answer for each question. Show all questions $\leq \Rightarrow$ The rate law for a particular reaction is determined to be $R = k[A]^2 [B]^2$.

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