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OBJECTIVES ; Describe

how to write a word

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OBJECTIVES

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CHAPTER 8 REVIEW
Chemical Equations
and Reactions MIXED
REVIEW SHORT

ANSWER Answer the
following questions in
the space provided. 1.

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b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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Chemical Equations &

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SHORT ANSWER

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Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid.

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CHAPTER 8 REVIEW
Chemical Equations
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3 SHORT ANSWER

Answer the following questions in the space provided. 1. List four

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3 SHORT ANSWER

Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid. Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg. 2. Consider the metals iron and silver, both listed in Table 3 on page 286 of the text. Which one

8 Chemical Equations and Reactions

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4. a. Use five to six data points from Appendix Table A-8 in the text to sketch the curve for water vapor's partial pressure versus temperature on the graph provided below.

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20 25 30 Temperature

CC) Do the data points lie on a straight line?

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Measurements and
Calculations CHAPTER

2 REVIEW

Measurements and

Calculations SECTION 1

SHORT ANSWER

Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model.

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Page 23/25

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Chapter 2 Section 3

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